

Contents lists available at ScienceDirect

## European Journal of Pharmaceutics and Biopharmaceutics

journal homepage: www.elsevier.com/locate/ejpb



## Comment on "Gelation of microemulsions and release behaviour of sodium salicylate from gelled microemulsions" [Eur. J. Pharm. Biopharm. 71 (2009) 297]

Cosima Stubenrauch a,\*, Thomas Sottmann b, Reinhard Strey b, Iseult Lynch c

The paper cited in the title [1] deals with the potential use of gelled microemulsions in drug delivery systems and the authors claim that they prepared a novel gelled microemulsion with which they measured release rates of a model drug. However, no experimental evidence is provided for the formation of microemulsions and we have severe doubts that a gelled microemulsion is formed under the experimental conditions described. The main points of criticism are as follows:

- (1) The authors do not consider that the phase behaviour and thus the microstructure of microemulsions strongly depend on the temperature and the composition [2]. Although measurements are carried out at different temperatures (FT-IR at RT, POM at 75 °C RT, release rates at 37 °C, sol–gel transitions at various *T*) and compositions, no phase studies are reported. In addition, the influence of the gelator and the model drug on the phase behaviour is not studied [3]. How did the authors locate the 1-phase microemulsion region?
- (2) The chosen gelator is unsuitable as the binary organogel is turbid [3], raising the question of how the authors distinguished between a gelled, phase separated system and a gelled 1-phase microemulsion.
- (3) The authors consider propylene glycol (PG) as co-surfactant referring to [4]. This, however, is incorrect. According to Table1 of [4] the droplet size does not change with increas-

- ing PG content, which can only be the case if PG is a co-solvent. Addition of a co-surfactant to a 1-phase microemulsion would increase the interfacial area and hence decrease the droplet size.
- (4) No experimental evidence is given for the structure proposed in Fig. 6. A bicontinuous structure cannot coexist with water droplets in an equilibrated microemulsion as this would require two different curvatures. Moreover, the different compositions of the samples automatically lead to different microstructures, i.e. no general picture can be drawn.

## References

- [1] G. Feng, Y. Xiong, H. Wang, Y. Yang, Eur. J. Pharm. Biopharm. 71 (2009) 297.
- [2] T. Sottmann, C. Stubenrauch, in: C. Stubenrauch (Ed.), Microemulsions: Background, New Concepts, Applications, Perspectives, Wiley-Blackwell, Oxford, 2009 (Chapter 1);
  - T. Sottmann, R. Strey, in: J. Lyklema (Ed.), Soft Colloids V Fundamentals in Interface and Colloid Science, Elsevier, Amsterdam, 2005 (Chapter 5).
- [3] C. Stubenrauch, R. Tessendorf, R. Strey, I. Lynch, K. Dawson, Langmuir 23 (2007) 7730:
  - C. Stubenrauch, R. Tessendorf, A. Salvati, D. Topgaard, T. Sottmann, R. Strey, I. Lynch, Langmuir 24 (2008) 8473.
- [4] H. Chen, D. Mou, D. Du, X. Chang, D. Zhu, J. Liu, H. Xu, X. Yang, Int. J. Pharm. 341 (2007) 78.

<sup>&</sup>lt;sup>a</sup> University College Dublin, School of Chemical & Bioprocess Engineering, Belfield, Dublin, Ireland

<sup>&</sup>lt;sup>b</sup> Institut für Physikalische Chemie, Universität zu Köln, Köln, Germany

<sup>&</sup>lt;sup>c</sup> University College Dublin, School of Chemistry & Chemical Biology, Belfield, Dublin, Ireland

<sup>\*</sup> Corresponding author. Address: University College Dublin, School of Chemical & Bioprocess Engineering, Belfield, Dublin 4, Ireland. Tel.: +353 1 7161923; fax: +353 1 7161177.